

## Discovering the Unexplored: Synthesis and Analysis of a New Orthorhombic Sn<sub>3</sub>O<sub>4</sub> Polymorph

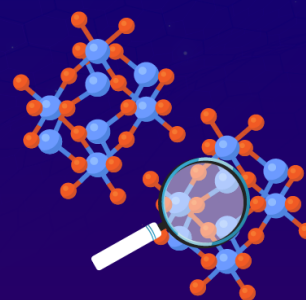
Tuning the reaction conditions such as degree of filling and gas composition can have a major impact on the products obtained by hydrothermal synthesis. This was clearly represented in the new Tokyo Tech study where they synthesized an unreported orthorhombic polymorph of Sn<sub>3</sub>O<sub>4</sub> instead of conventional monoclinic phase by optimizing the conditions inside the hydrothermal reactor. The orthorhombic Sn<sub>3</sub>O<sub>4</sub> has a narrower bandgap than the conventional one, thus making it useful as a visible-light active photocatalyst.

### New Strategy for Fabrication and Analysis of Unexplored Sn<sub>3</sub>O<sub>4</sub> Phase

Tin oxides (Sn<sub>x</sub>O<sub>y</sub>) exhibit a wide range of unique optical and electrical properties due to their multiple oxidation states

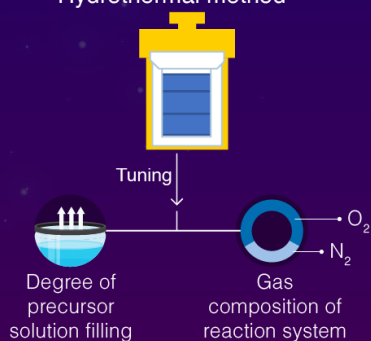


Exploring newer tin oxides with different chemical formulas and phases can help expand their applications further



### Synthesis of a new phase of Sn<sub>3</sub>O<sub>4</sub> polymorph

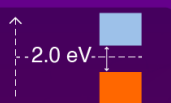
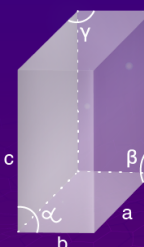
Hydrothermal method



New Sn<sub>3</sub>O<sub>4</sub> polymorph

- Formula: Sn(II)<sub>2</sub>Sn(IV)O<sub>4</sub>
- Crystal phase: Orthorhombic

$$a \neq b \neq c$$
$$\alpha = \beta = \gamma = 90^\circ$$



✓ Small bandgap (2.0 eV)



✓ Wide range of visible-light absorption

**The synthesis of Sn(II)<sub>2</sub>Sn(IV)O<sub>4</sub> provides insights on the optimization of hydrothermal synthesis methods and facilitates the discovery of new oxide materials**

Synthesis and Characterization of the Orthorhombic Sn<sub>3</sub>O<sub>4</sub> Polymorph  
Miyauchi et al. (2023) | *Angewandte Chemie International Edition*



Oxides of tin (Sn<sub>x</sub>O<sub>y</sub>) are found in many of modern technologies due to their versatile nature. The multivalent oxidation states of tin—Sn<sup>2+</sup> and Sn<sup>4+</sup>—impart tin oxides with electroconductivity, photocatalysis, and various functional properties. For the photocatalysis application of tin oxides, a narrow bandgap for visible-light absorption is indispensable to utilize a wide range of solar energy. Hence, the discovery of new Sn<sub>x</sub>O<sub>y</sub> could help improve

the efficiency of many environmentally significant photocatalytic reactions like water splitting and CO<sub>2</sub> reduction. While there are many theoretical and computational predictions of new stable Sn<sub>x</sub>O<sub>y</sub>, there still remains a need for experimental studies that can turn the predictions into reality.

Taking this as a call to action, researchers from Tokyo Institute of Technology, National Defense Academy, and Mitsubishi Materials Corporation have designed a new tin oxide. In their recent breakthrough published in *Angewandte Chemie International Edition*, Mr. Y. Liu et al. presented a new optimized hydrothermal synthesis approach that led to the synthesis of a Sn<sub>3</sub>O<sub>4</sub> polymorph with a previously unreported orthorhombic crystal structure. The research was performed in the Mitsubishi Materials Sustainability Innovation Collaborative Research Cluster with the support of the Tokyo Institute of Technology Open Innovation Platform.

The project leader, Prof. Miyauchi explains the driving force behind the study, “The aim of our study was two-fold. First was the synthesis of a new tin oxide polymorph and the second was applying it for a visible-light sensitive photocatalyst.”

The team set up multiple thermal hydrothermal reactors with the same starting material for preparing Sn<sub>3</sub>O<sub>4</sub>. In the first series one set, they altered the degree of filling of the precursor solution by filling 20, 40, 60, and 80% of a 100 ml Teflon liner. For the second series, they kept the degree of filling constant at 20% and the Teflon liners were filled with ambient air, pure oxygen, and pure nitrogen respectively.

The team then carried out Rietveld analysis, X-ray spectroscopy, and first-principles calculations on the products formed. The analysis showed the new Sn<sub>3</sub>O<sub>4</sub> polymorph has the chemical formula of Sn(II)<sub>2</sub>Sn(IV)O<sub>4</sub>. Its X-ray diffraction pattern has never been reported and is assigned to an orthorhombic crystal phase based on empirical and computational analyses. The comparative studies for tuning of gas composition and degree of filling showed that the orthorhombic polymorph was only formed when the degree of filling was high or when the gas introduced was inert and has less oxygen. The team hence suggested that paying attention to the oxygen source could be the key to more precise hydrothermal synthesis.

The novel orthorhombic Sn<sub>3</sub>O<sub>4</sub> polymorph reported in this study has a smaller bandgap than a conventional monoclinic Sn<sub>3</sub>O<sub>4</sub>, indicating a higher efficiency of absorbing visible light. Furthermore, the conduction band of the orthorhombic polymorph is enough high to drive CO<sub>2</sub> reduction reaction.

Hydrothermal method is a widely used method of materials synthesis. This study finds that the often-neglected parameters in hydrothermal synthesis drastically affect the crystal structure. This finding is informative for the discovery of numerous new oxide materials.

## Reference

Authors: Yang-Shin Liu,<sup>1</sup> Akira Yamaguchi,<sup>1</sup> Yue Yang,<sup>1</sup> An Niza El Aisnada,<sup>1</sup> Sho Uchida,<sup>1</sup> Hideki Abe,<sup>2</sup> Shigenori Ueda,<sup>3,4,5</sup> Kenji Yamaguchi,<sup>6</sup> Toyokazu Tanabe,<sup>7</sup> and Masahiro Miyauchi\*<sup>1</sup>

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Affiliations: <sup>1</sup> Department of Materials Science and Engineering, School of Materials and Chemical Technology, Tokyo Institute of Technology, Japan  
<sup>2</sup> Center for Green Research on Energy and Environmental Materials, National Institute for Materials Science, Japan.  
<sup>3</sup> Synchrotron X-ray Station at SPring-8, National Institute for Materials Science, Japan.  
<sup>4</sup> Research Center for Advanced Measurement and Characterization, National Institute for Materials Science, Japan.  
<sup>5</sup> Research Center for Functional Materials, National Institute for Materials Science, Japan.  
<sup>6</sup> Innovation Center, Mitsubishi Materials Corporation, Japan.  
<sup>7</sup> Department of Materials Science and Engineering, National Defense Academy, Japan.

\*Corresponding author's email: [mmiyauchi@ceram.titech.ac.jp](mailto:mmiyauchi@ceram.titech.ac.jp)

## Further information

Professor Masahiro Miyauchi

Department of Materials Science and Engineering

School of Materials and Chemical Technology

Tokyo Institute of Technology, Japan

[mmiyauchi@ceram.titech.ac.jp](mailto:mmiyauchi@ceram.titech.ac.jp)